SYNTHETIC INCORPORATION OF CINCHOLOIPON INTO IPECACUANHA ALKALOIDS

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Continuing interest in syntheses of the ipecacuanha and analogous alkaloids prompts us to record the title incorporation accomplished in our laboratory. Although the extensive effort by many groups toward a total synthesis of the principal alkaloid emetine (1) had culminated in at least a dozen suc**cessful achievements,' the present synthesis was undertaken as the vehicle for synthetic efforts to**ward general preparations of these alkaloids from the major cinchona alkaloids (4) through the ethyl **ester (?) of cincholoipon where the later reaction steps could take advantage of high stereoselectivity due to the two asymmetric centers' at C-3 and C-4 and of omitting optical resolution.**

Of the various strategies conceivable for transformation of cis-ester 2 into an appropriat known key intermediate leading to J and related alkaloids, the one we selected required four main operations: (i) introduction of the 3,4-dimethoxyphenethyl skeleton into 2 at N-1; (ii) construction of the **lactam carbonyl function at C-6; (iii) epimerization at C-4 to form the 3,4-trans configuration which** should correspond to the relative and absolute configuration of 1, at the 2- and 3-positions; (iv) ring closure to complete the benzoquinolizidine part (24) . To embody the second operation by the mercuric acetate--(ethylenedinitrilo)tetraacetic acid (EDTA) oxidation method,^{3,4} the incoming carbon skeleton **in the first operation should carry a hydroxyl group or its equivalent at the benzylic position. Thus,** treatment of $(+)$ -2, prepared in 31% overall yield from cinchonine (5) by the known method,^{22,5} with **3,4-dimethoxyphenacyl bromide and K2CO3 in benzene furnished** $(+)-6$ **[89% yield;** $[\alpha]_D^{24}$ + 2.3° (c 2.5, EtOH)],⁶ which was reduced (NaBH4, EtOH, 0-5[°], 4 hr) to yield a diastereoisomeric mixture of $\frac{7}{4}$ in **90% yield.**

Oxidation of the mixture (7) with Hg(OAc)₂—EDTA (1% aqueous AcOH, reflux, 1.5 hr)^{3,4} and hydrogenolysis of the resulting lactamalcohols (Pd-C/H₂, EtOH-70% HClO₄, 20[°], 4.2 atm, 24 hr) gave $(-)-8$ $[44\%; [\alpha]_D^{18} -8.4^{\circ} (\alpha)$ **EtOH**], $(-)-11$ $[4\%; [\alpha]_D^{25} -3.5^{\circ} (\alpha)$ **EtOH**], and $(+)-13$ $[30\%; [\alpha]_D^{25}$

2, $R = CO₂Et$ $3, R = CH₂OH$

 A , R=H or OMe; R'=vinyl or Et λ , R=H; R'=vinyl

6, $R = CO₂Et$; $Y = H₂$; $Z = O$ Z , R = CO2Et; Y = H2; Z = H, OH $8, R = CO₂Et; Y = O; Z = H₂$ 9, $R = CO₂H$; $Y = O$; $Z = H₂$ (±)-10, R = CO2Me; Y = O; Z = O

11, $R = CO₂Et$; $Y = O$ 12, $R = CH₂OH$; $Y = H₂$

13, $R = CO₂Et$; $Y = O$ 14 , R = CH₂OH; Y = H₂

 $15, R = CO₂H$; Y = O $16, R = CO₂Et; Y = O$ 17, $R = CH_2OH$; $Y = H_2$

 (\pm) -18, R = CO2H (\pm) -19, R = CO2Me

 $(\pm) - 20$

 \cdot

 $\frac{23}{2}$

 (\pm) -21, R = CO2Et; Y = O (\pm) -22, R = CH2OH; Y = H2 +26.1[°] (c₁, EtOH). The structure of (-)-8 was confirmed by hydrolysis (1 N NaOH-EtOH, 20[°]) leading to $(-)-2$ [96%; $[\alpha]_0^{20} -1.4^{\circ}$ (c_1, EtOH)] and spectral comparison of the acid with $(\pm)-2$ synthesized by the following reaction sequence. Esterification of (\pm) -18 (mp 203-204[°])⁷ with diazomethane and treatment of the resulting (\pm) -19 with dimethyl sulfate (benzene, reflux, 3 hr) produced (\pm) -20 [60% overall yield; bp 97[°] (2 mm)], which reacted with 3,4-dimethoxyphenacyl bromide (60[°], 5 hr) to provide (±)-10 (84%; mp 122-123[°]).⁸ Conversion of (±)-10 into (±)-2⁹ (90% overall yield; mp 150-152°) was effected by the NaBH4 reduction followed by hydrogenolysis (Pd-C/H2) and alkaline hydrolysis. On the other hand, the LiAlH4 reduction of $(-)$ -11 in ether afforded $(+)$ -12,¹⁰ identical with a sample $[(\alpha]_D^{30} + 9.1^\circ \ (\underline{c} \ 2.4, \text{ EtoH})]$ obtained by the reaction of $(+)-\frac{3}{2}$ ² with 3,4-dimethoxyphenethyl bromide (KzCO₃, benzene, reflux). Likewise, the reduction of $(+)$ -13 gave $(-)$ -14 $[(\alpha]_D^{30}$ -41.6° (c_ 2, EtOH)] and its 3,4-trans structure was established by spectral comparison with (1) - $14'$ ¹⁰ derived from (\pm) -21 (mp 93-94[°])⁷ through (\pm) -22. The occurrence of $(+)$ -13 is probably due to epimerization of $(-)-11$ at C-3, hence in the Hg(OAc)z-EDTA oxidation of $\bar{\lambda}$ the ratio of the 6- to the 2-oxidation **may be compared with that" observed for simpler 3-alkylpiperidine derivatives.**

Isomerization of $(-)$ -cis-acid \int to $(+)$ -trans-acid \int , $[mp 129-130^\circ; [\alpha]_D^{26} + 63^\circ (\underline{c} 1, EtoH);$ solution ir and nmr spectra identical with those of authentic $(\pm)-15$ ¹²] was carried out in boiling 10% hydrochloric acid or, more efficiently, without the solvent at 180° for 80 min, resulting in an equilibrated mixture $(2:15=33:67)$. Yield of $(+)$ -15 could be raised to 83% when the cis-acid recovered **from the mixture was repeatedly subjected to the same reaction. The isomerization appears most probably to proceed through hydrolytic ring opening and recyclization'3 or through intramolecular participation of the exocyclic carboxyl group whereby the side chain at C-4 exchanges places with** the endocyclic C-5-C-6 chain. Esterification of $(+)$ -15 (EtOH--HCl, 25[°]) furnished $(+)$ -16 [92%; $[\alpha]_D^{28}$ +54.3^o (c 1, EtOH)], which gave (+)-17 [96%; $[\alpha]_D^{30}$ +40.4^o (c 2.1, EtOH)], the enantiomer of **(-)-2, on reduction with LiAlH4.**

Compound $(+)$ -16 was cyclized (POC1₃, toluene, reflux, 1.5 hr) to $22(X=1)$ (95%; mp 153-154[°]). Conversion of the iodide salt into 23 (X=ClO₄) (mp 133-134[°]) and subsequent catalytic hydrogenation $(PtO₂, EtOH, 20^o)$ produced $24 \cdot HClO₄$ (mp 149-150[°]), from which base $(-)-24$ [98% overall yield; mp $90-91^\circ$; $\left[\alpha\right]_D^{30}$ -39.3° (c_ 1, EtOH)] was obtained.¹⁴ The physical properties of this sample, in good agreement with those reported,¹⁵ and spectral comparison with authentic (\pm) -24¹² unequivocally established its structure. Since tricycle $(-)$ -24 has been shown to lead to Q -methylpsychotrine,¹⁵ emetine (1),^{'5b} psychotrine,¹⁶ protoemetine,¹⁷ and tubulosine alkaloids,¹⁸ the preparation of this substance formally concluded syntheses of these alkaloids. (-)-c<u>is</u>-Acid y should give an enantiomer of

cis-emetine by following the same route as reported for the (\pm) -series.⁹⁴

The synthetic scheme described may serve as a valid model for the preparation of other analogous alkaloids ' **which still remain to be synthesized.**

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REFERENCES

- **1. For recent reviews, see (a) A. Brossi,** S . **Teitel, and G . V. Parry in "The Alkaloids ," Vol. XIII, R. H. F. Manske, Ed., Academic Press, New York, N. Y., 1971, Chapter 3; (b) H. T. Openshaw in "Chemistry of the AIkaloids ,"** S . **W. Pelletier, Ed., Van Nostrand Reinhold Co., New York, N. Y., 1970, Chapter 4.**
- 2. For the configuration, see (a) V. Prelog and E. Zalan, Helv. Chim. Acta, 27, 535 (1944); (b) idem, ibid., 27, 545 (1944); (c) W. Solomon in Reference 1b, Chapter 11.
- 3. H. Möhrle, Arch. Pharm., 297, 474 (1964).
- 4. T. Fujii and S. Yoshifuji, Chem. Pharm. Bull. (Tokyo), 20, 1451 (1972).
- 5. A. Kaufmann, E. Rothlin, and P. Brunschweiler, Ber., 49, 2299 (1916).
- **6. Satisfactory spectral data and/or elemental analyses were obtained for all new compounds.**
- **7. (a) T. Fujii, Chem. Pharm. Bull. (Tokyo), 2, 591 (1958); (b) R. J. Sundberg and F. 0. Holcombe, Jr., J. Org. Chem., 2, 3273 (1969); (c) T. Fujii, S. Yoshifuji, and M. Tai, Abstracts of Papers, 36th Meeting of Hokuriku Branch, Pharmaceutical Society of Japan, Kanazawa, Japan, June, 1973, p. 5.**
- 8. For the general utility of this reaction, see T. Fujii, S. Yoshifuji, and K. Yamada, Chem. Ind. **(London), in press.**
- **9. (a) M. Barash, J. M. Osbond, and J. C. Wickens, J. Chem. Sot., 3530 (1959); (b) E. E. van** Tamelen, P. E. Aldrich, and J. B. Hester, Jr., J. Am. Chem. Soc., 81, 6214 (1959).
- **10. A. Brossi, A. Cohen, J. M. Osbond, P. Plattner, 0. Schnider, and J. C. Wickens, J. Chem. Sz., 3630 (1959).**
- 11. **T. Fujii, S. Yoshifuji, K. Michishita,** M. **Mitsukuchi, and K. Yoshida, Chem. Pharm. Bull. (Tokyo), 2, 2695 (1973).**
- **12. A. R. Battersby and J. C. Turner, J. Chem. Sot., 717 (1960).**
- **13.** (a) T. Fujii and S. Yoshifuji, Tetrahedron, 26, 5953 (1970); (b) idem, Chem. Pharm. Bull. **(Tokyo), 2, 1051 (1971); (c) T. Fujii, S. Yoshifuji, and A. Tamai, ibid., 2, 369 (1971).**
- 14. In the (\pm) -series, the conversion of lactam 15 to tricycle 24 or its methyl ester analog has been **reported in References 9a, 9b, and 12.**
- **15. (a) A. R. Battersby and B. J. T. Harper, J. Chem. SOC., 1748 (1959); (b) H. T. Openshaw** and N. Whittaker, ibid., 1461 (1963).
- 16. S. Teitel and A. Brossi, J. Am. Chem. Soc., 88, 4068 (1966).
- **17. C. Sz&ntay, L. Take, and P. Kolonits, J. Org. Chem.,** 2, **1447 (1966).**
- **18. H. T. Openshaw and N. Whittaker, J. Chem. Sot. (C), 91 (1969).**